UNIT 7: DISTRIBUTION



CORAL REEF ECOLOGY CURRICULUM



This unit is part of the *Coral Reef Ecology Curriculum* that was developed by the Education Department of the Khaled bin Sultan Living Oceans Foundation. It has been designed for secondary school students, but can be adapted for other uses. The entire curriculum can be found online at *lof.org/CoralReefCurriculum*.

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STANDARDS

- <u>CCSS</u>: RST.9-10.1, 2, 3, 4,
 5, 7, 8, 10; RST.11-12.1, 2, 3,
 4, 8, 10; SL.9-10.1, 6; SL.11 12.1, 6; HSN.Q.A.1; HSA.
 CED.A.1, 4
- <u>NGSS</u>: ESS 2.A, ESS 2.C, ESS 2.D, HS-LS2-2, HS-LS2-6
- **OLP**: 1.B.1, 1.C.1, 1.C.7, 1.C.8, 1.C.9, 1.C.11

ONLINE CONTENTS

- <u>Distribution Quiz</u>
- <u>Where Are Coral Reefs</u> <u>Found? Video</u> Although corals are found throughout the planet, most reef-building corals are found in the tropics and subtropics where thousands of animals make these reefs their home.

DISTRIBUTION

This lesson is a part of the *Distribution* unit, which explains the two major drivers of coral distribution: salinity and temperature. Below is a summary of what is included in the entire unit.

UNIT CONTENTS

A. Background Information

- Where are Corals Found?
- What is a Current?
- What is Density?
- Salinity, Temperature, and Ocean Circulation
- B. Lessons

Watch it! Where are Corals Found?

 A worksheet to accompany the <u>Where are Corals Found?</u> video

Density 101

A lab to calculate and compare densities of liquids

Inquiring about Density 1

• A lab to create a procedure to determine relative densities

Inquiring about Density 2

• A lab to create a procedure to determine actual densities

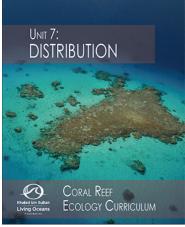
Go With the Flow

• A worksheet to accompany a teacher demonstration on how salinity and temperature affect water density

Read it! Galapagos Ocean Currents

• A worksheet to accompany the <u>Galapagos Ocean Currents</u> field blog





A) WHERE ARE CORALS FOUND?

Distribution is the geographic area in which a species can be found. In this unit, we will learn about the two major drivers of coral distribution.

Most people think that corals can only grow in warm subtropical and tropical waters; however, they are mistaken. Corals can also grow in higher latitudes and greater depths. These corals are often referred to as **deep-** or **cold-water corals**. They do not share the same environmental conditions as subtropical and tropical corals, which are referred to as **warm-water corals**. For now, we will focus on warm-water coral reefs and in *Unit 8* we will learn about the environmental conditions that these reefs require for survival.

The **tropics** is a geographic region that is located between the Tropic of Cancer in the northern hemisphere (23.4378° N) and the Tropic of Capricorn in the southern hemisphere (23.4378° S) (figure 7-1). The **subtropics** extend from the Tropics of Cancer and Capricorn to 35° North and South latitudes.

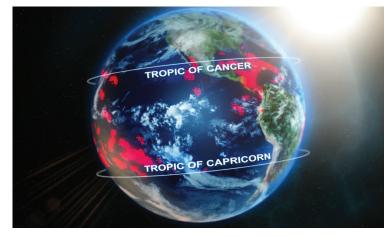


FIGURE 7-1. Most coral reefs (in red) are located between the Tropic of Cancer and the Tropic of Capricorn.

Typically, coral reefs can be found in the tropics. There are some areas outside this range (in the subtropics) where coral reefs exist due to currents transporting warm water, such as parts of the Bahamas, Bermuda, Florida, Gulf of Mexico, and Japan.

B) WHAT IS A CURRENT?

An ocean current is the continuous movement of ocean water in a prevailing direction.

There are two main types of ocean currents: deep ocean currents and surface currents (figure 7-2). **Deep ocean currents** are currents that occur greater than 300 feet (100 meters) deep. They are driven by density, temperature, and salinity. **Surface currents** are currents driven primarily by wind, tides, and waves.

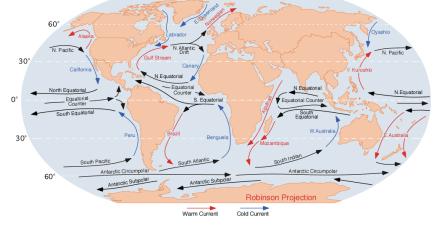


FIGURE 7-2.

Major ocean currents - the red arrows represent where warm currents are found and the blue arrows represent cold currents. In this unit, we will investigate how deep ocean currents work. In order to understand this type of current, we first have to understand the terms density, salinity, and temperature.

C) WHAT IS DENSITY?

Density is a physical property of **matter**, a substance that occupies space. The definition of density is the concentration of matter of an object. Density may also refer to how closely *packed* or *crowded* the material appears to be. Density is the same for a particular substance when measured at the same temperature and pressure, regardless of the quantity or shape of the sample. For example, a copper penny and a copper bracelet are going to have the same density despite the fact that the bracelet has a greater amount of copper.

The units of density are commonly expressed as grams per centimeter cubed (g/cm³) for solids, grams per milliliter (g/mL) for liquids, and grams per liter (g/L) for gases.

Think about density in terms of atoms that make up an object. For instance, let's take a look at these cubes (figure 7-3). One cube clearly has more atoms (red circles) than the other. **Atoms** are the most basic unit of chemistry and they have different number of protons, neutrons, and electrons. Which cube do you think has a higher density? That's correct! The cube with more atoms has a greater density because they are more closely *packed*. When the volume is the same and there is a greater mass, the density is greater. So we can say that the first cube has a greater density than the second.

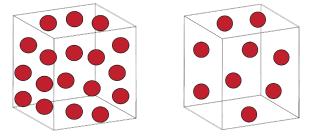


FIGURE 7-3. Comparing density of cubes with the same volume

We can actually prove this mathematically too. In order to find the density, we use the equation:

Density = $\underline{\text{mass}}$ or $D = \underline{m}$ Volume V

Mass: The amount of matter an object contains. The mass of a substance is determined by the mass of the atoms, **molecules**, or **ions** that make up the substance (see *Glossary*).

Volume: A measure of the amount of space an object takes up.

By taking the object's mass divided by its volume, we can find the density of the object.

Say that the cubes in figure 7-3 have a volume of 64 cm³ and each atom has a mass of 4.0 grams. We can calculate the density of each cube.

	Cube 1	Cube 2
Volume	64 cm ³	64 cm ³
Mass	18 atoms x 4.0 grams per atom = 72 grams	9.0 atoms x 4.0 grams per atom = 36 grams
D= <u>m</u> V	72 g/64 cm ³	36 g/64 cm ³
Density	1.125 g/cm ³	0.5625 g/cm ³

The math proves that the density of the first cube is greater than the density of the second.



D) SALINITY, TEMPERATURE, AND OCEAN CIRCULATION

We just learned about density. Seawater density drives deep ocean currents. Now we are going to learn about two factors that affect the density of sea water: salinity and temperature.

If you've been swimming at the beach, you know that the ocean consists of salt water. The salt in seawater is the same thing that we add to our food. Chemically, salt is a **compound** because it is made up of more than one element; it is known as sodium chloride (NaCl) (figure 7-4).

How does this relate to salinity? **Salinity** is the amount of salt dissolved in a body of water. Throughout the ocean and different bodies of water, the

amount of salt fluctuates depending on factors such as evaporation, rainfall, FIGURE 7-4. Salt (sodium runoff, and ice formations.



chloride)

We measure salinity in parts per thousand (ppt). The average ocean salinity is 35 ppt. Let's think of salinity like this: if we have 1,000 buckets, 35 of those buckets will be filled with salt and the other 965 will contain fresh water.

In areas where there is a greater addition of fresh water (due to rainfall or runoff), there will be lower salinity and lower density because the fresh water will dilute the salt water. When there are more ice formations or evaporation occurs, there is a higher salinity and higher density because there is less fresh water to dilute the salt water.

OCEAN TEMPERATURE

Just like salinity, the temperature of the ocean also fluctuates. **Temperature** is the measure of how hot or cold something is. In physics, it's defined as the measure of the kinetic energy of particles. It can be measured in Fahrenheit, Celsius, or Kelvin scale.

The ocean has three temperature layers. Study figure 7-5 to visualize the changes in temperature and depth.

- 1. Surface Layer: Often this layer is referred to as the mixed layer because this is where winds and waves stir up the water. This layer is also highly influenced by solar energy (from the sun) heating up the water. The temperature is roughly that of the air above it.
- 2. Thermocline: A layer of water located between the surface and deep water layers that rapidly decreases in temperature as depth increases.
- 3. Deep Water Layer: The layer of water below the thermocline that slowly decreases in temperature as depth increases.

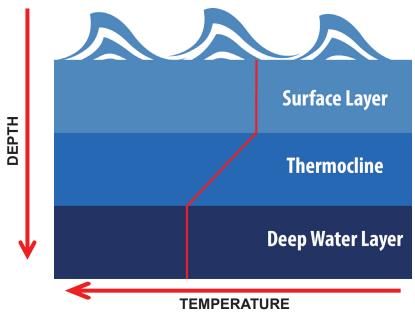


FIGURE 7-5.

Temperature layers in the ocean - as depth increases, temperature decreases: there is a sharp change in temperature only in the thermocline.

Due to different climates, the depths of these ocean layers fluctuate throughout the world. The largest of these currents is referred to as the **global ocean conveyor belt**. It is more formally called **thermohaline circulation**. Let's break down the word thermohaline.

thermo haline temperature salinity

Thermohaline currents are driven by temperature and salinity.

Temperature and density have an *inverse relationship*, which means as temperature increases, density decreases, and vice versa. Think about this at a molecular level. As the temperature of water increases, the water molecules have more energy and move faster, causing the distances between them to begin to increase. This expansion causes the molecules to spread out and become less dense. As the temperature of the water decreases, the water molecules slow down and tightly pack together, causing them to become more dense.

Salinity and density have a *direct relationship*, which means as salinity increases, density increases, and vice versa. Again, think about this on a molecular level. Salt is composed of sodium ions (Na⁺) and chloride ions (Cl⁻). Each of these ions has mass. As we increase the number of ions, the mass also increases, causing the density to increase.

Now let's apply what we just learned about temperature and salinity to thermohaline circulation (figure 7-6). Cold, salty water is more dense than warm, salty water and sinks. Warm, salty water is less dense and rises to the surface. This current starts where the Gulf Stream meets the Norwegian Sea. The Gulf Stream transports warm water north that has been heated in the lower latitudes. When this warm water gets to the colder climate in the higher latitudes of the Norwegian Sea, it cools and sinks. This bottom water will flow south and eventually make its way to Antarctica. Then the processes of warming and cooling occurs all over again. In time, this water will circulate the globe.

Scientists believe that it takes 1,000 years for the global ocean conveyor belt to make one cycle!

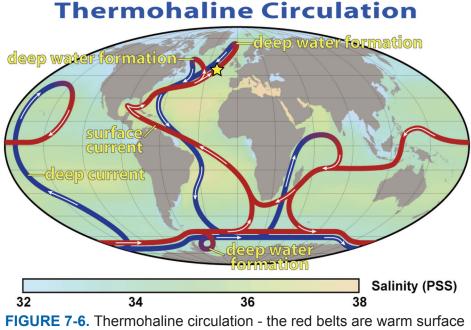


FIGURE 7-6. Thermohaline circulation - the red belts are warm surface currents and the blue belts are deep cold currents. The star marks the area where the Gulf Stream and Norwegian Sea meet.

Coral distribution is largely dependent on temperature, but now we know that both salinity and temperature affect ocean currents. Currents allow corals to have a wider distribution, providing warm water to the subtropics.



Figure 7-2. By L30nc1t0 [Public domain], 10 August 2007 via Wikimedia Commons. <u>http://commons.wikimedia.</u> org/wiki/File%3ACorrientes-oceanicas.gif.

Figure 7-4. By Mark Schellhase (Own work) [CC-BY-SA-3.0 (<u>http://creativecommons.org/licenses/by-sa/3.0</u>) or GFDL (<u>http://www.gnu.org/copyleft/fdl.html</u>)], 25 July 2008 via Wikimedia Commons. <u>http://commons.</u> wikimedia.org/wiki/File%3ASalt_Crystals.JPG.

Figure 7-5. Adapted from University of Illinois. 2010. Retrieved February 10, 2014 from <u>ww2010.atmos.uiuc.</u> <u>edu/(Gh)/guides/mtr/eln/upw.rxml</u>.

Figure 7-6. By Robert Simmon, NASA. Minor modifications by Robert A. Rohde also released to the public domain (NASA Earth Observatory) [Public domain], 29 March 2008 via Wikimedia Commons. <u>http://commons.</u> <u>wikimedia.org/wiki/File%3AThermohaline_Circulation_2.png</u>.





CORAL REEF ECOLOGY CURRICULUM

The Coral Reef Ecology Curriculum is a comprehensive educational resource designed to educate people about life on coral reefs. Developed by educators and scientists at the Khaled bin Sultan Living Oceans Foundation, this curriculum strives to increase ocean literacy by creating awareness about coral reefs, the threats they face, and how people can help to preserve these diverse ecosystems.



The Khaled bin Sultan Living Oceans Foundation is a US-based nonprofit environmental science organization. The Foundation was established to protect and restore the world's oceans through scientific research, outreach, and education.